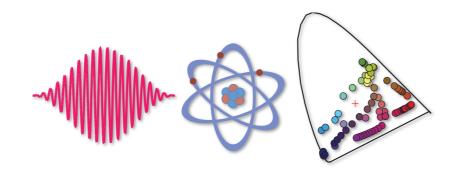
# Changing the color of atoms and molecules with light

#### Umberto De Giovannini

University of the Basque Country, San Sebastian, Spain

 $\underline{umberto.degiovannini@ehu.es}$ 





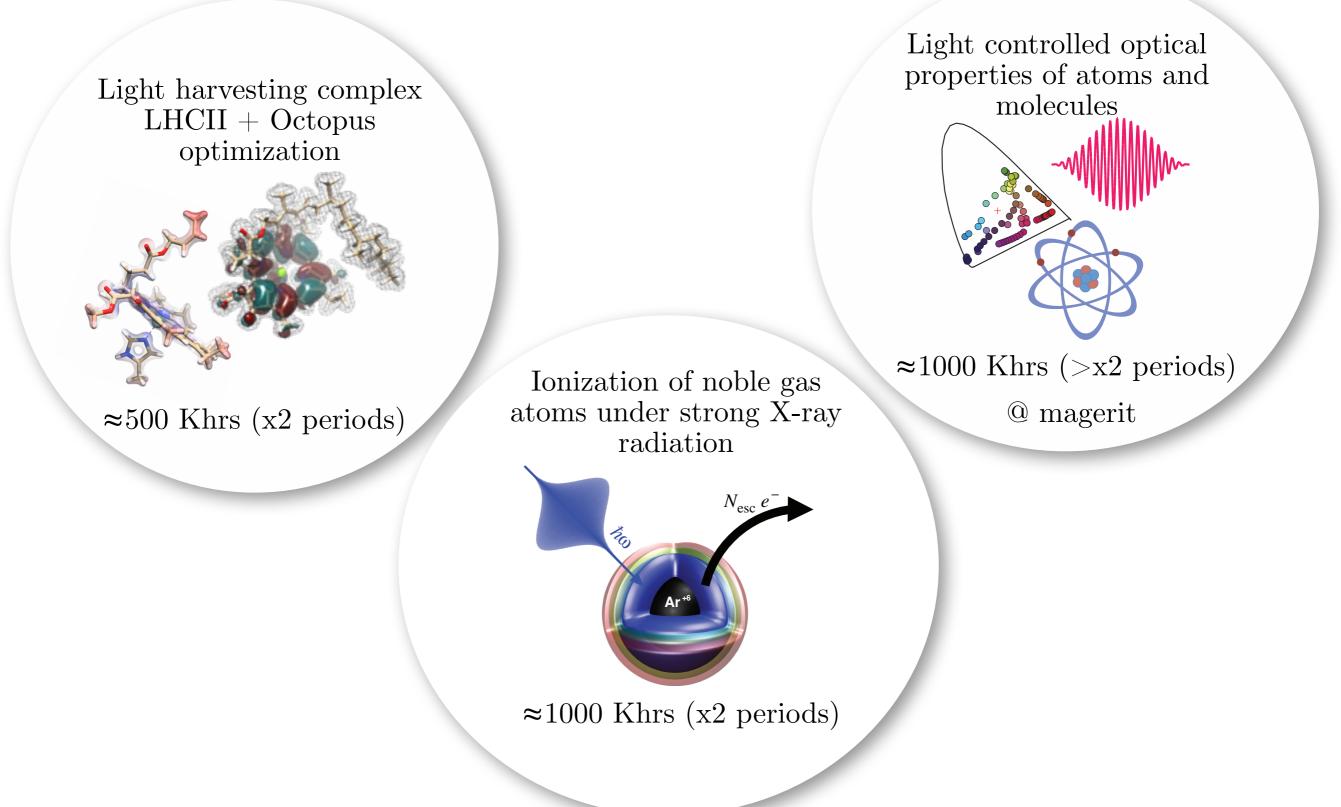


Universidad Euskal Herriko del País Vasco Unibertsitatea

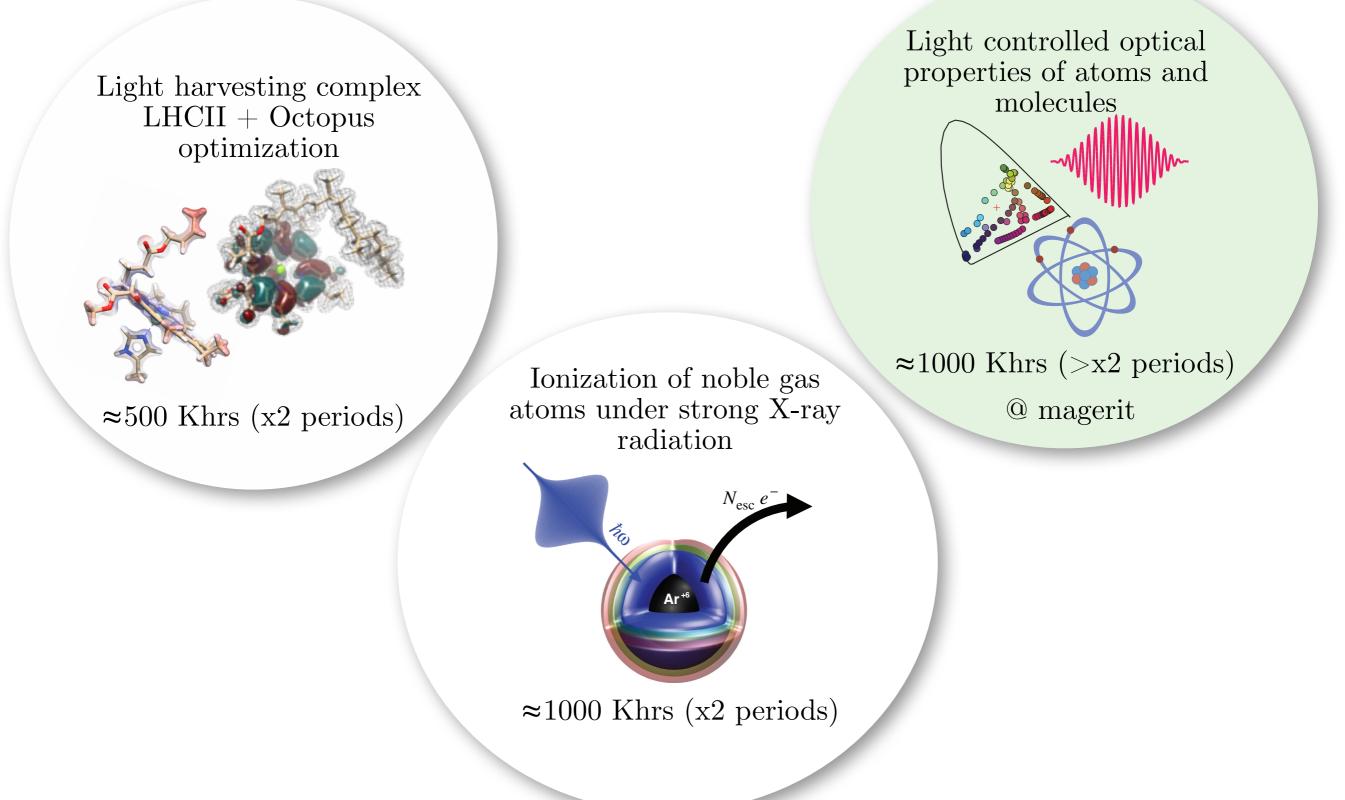


European Theoretical Spectroscopy Facility









#### People

Jessica Walkenhorst

Nano-Bio Spectroscopy Group and ETSF Scientific Development Center University of the Basque Country UPV/EHU Avenida de Tolosa 72, 20018 San Sebastian (Spain)





#### Alberto Castro

ARAID Foundation - Institute for Biocomputation and Physics of Complex Systems, University of Zaragoza Mariano Esquillor Gómez s/n, 50018 Zaragoza, (Spain)

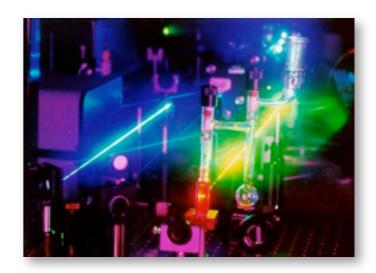
#### Angel Rubio

Nano-Bio Spectroscopy Group and ETSF Scientific Development Center University of the Basque Country UPV/EHU Avenida de Tolosa 72, 20018 San Sebastian (Spain)





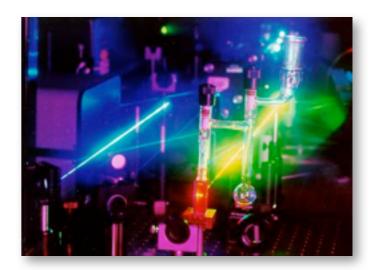
#### Motivation



#### To which extent it is possible to control the optical properties of matter with light?



#### Motivation



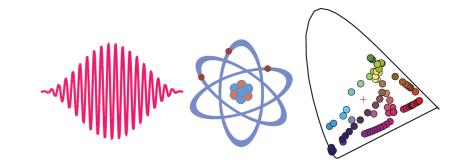
# To which extent it is possible to control the optical properties of matter with light?

#### A specific case:

Show how it possible to turn visible transparent gases of atoms and molecules and control their color with lasers



#### Outline



• Overview:

color perception idea

• Theory and tools: TDDFT OCT

• Applications: hydrogen and methane

• Conclusions













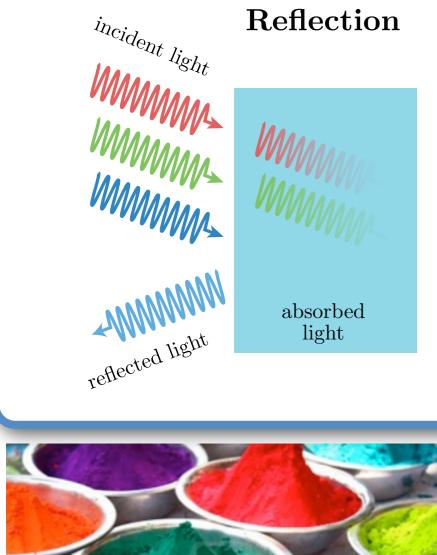


incident light

₩₩₩₩

₩₩₩₩

₩₩₩₩







Transmission

₩₩₩₩

absorbed

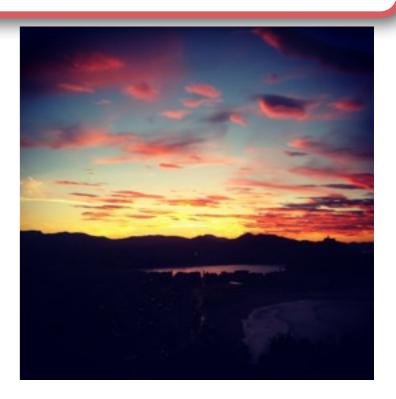
light

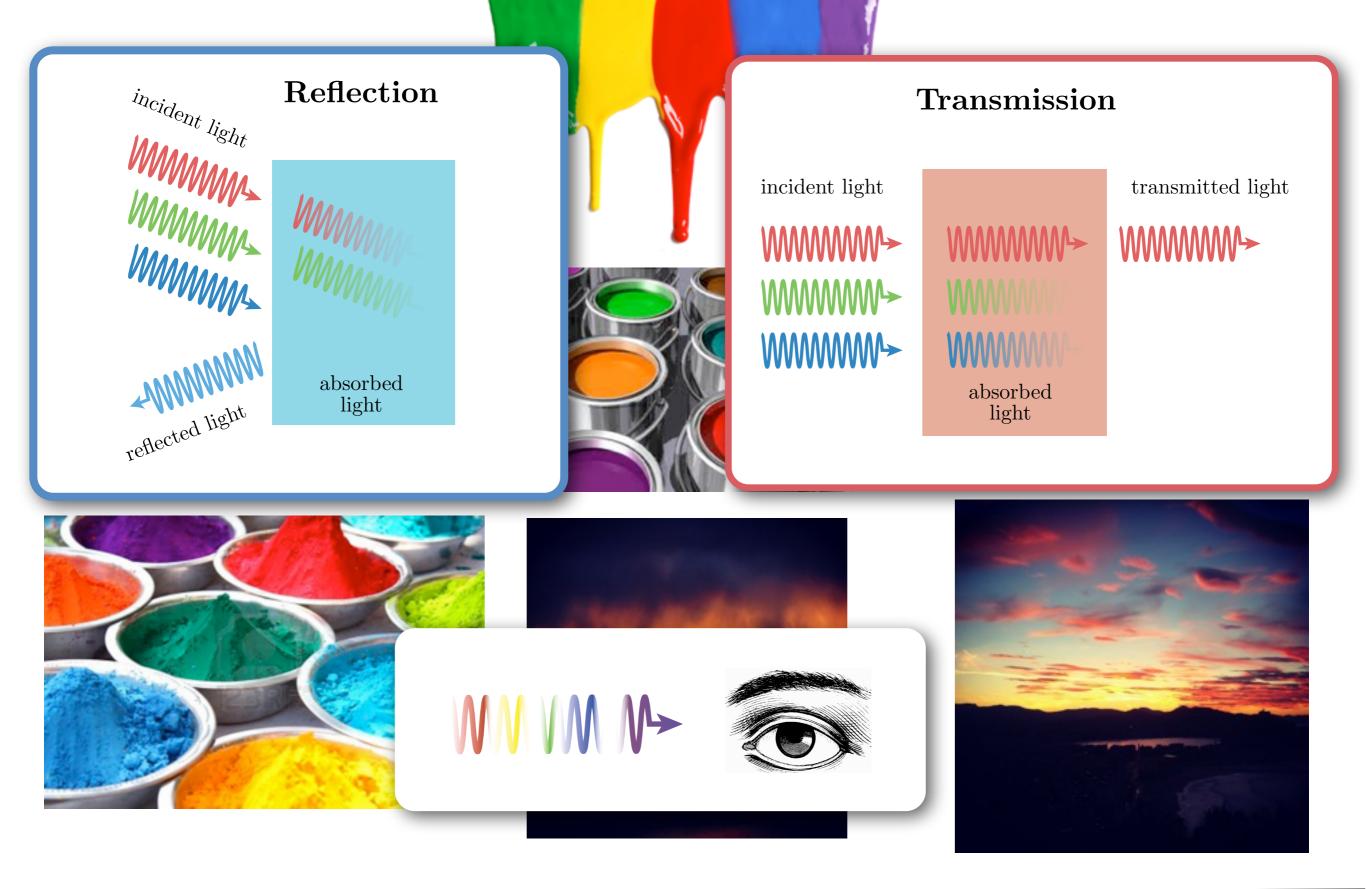
transmitted light

#### ₩₩₩₩

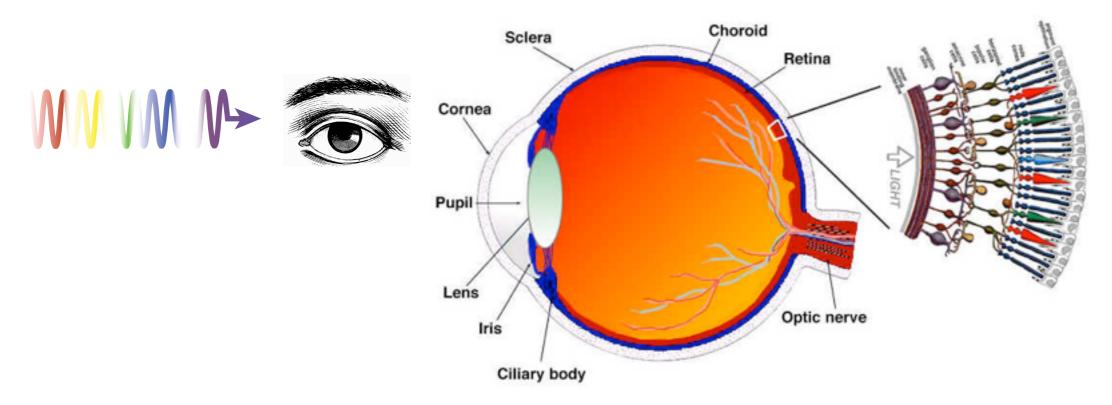


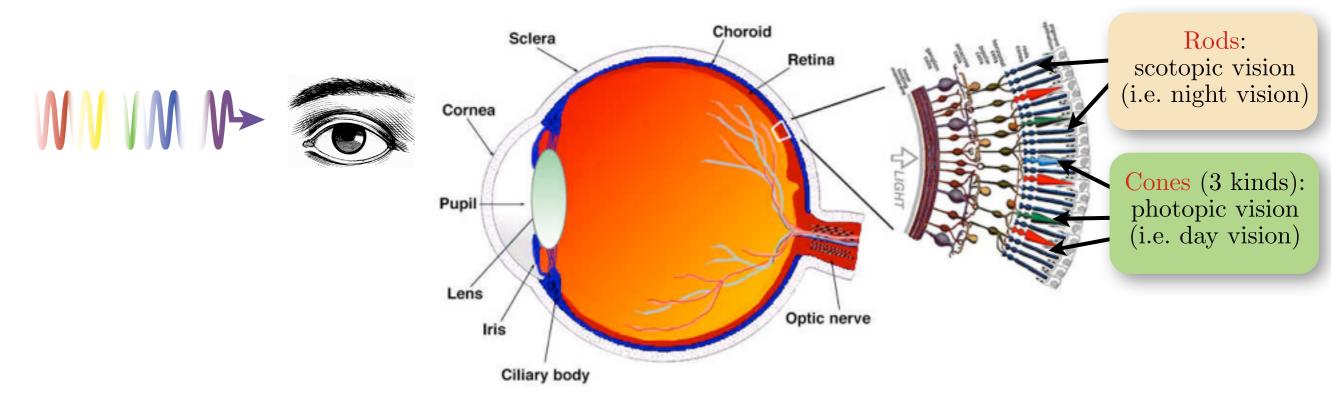
Santander September 23rd 2014

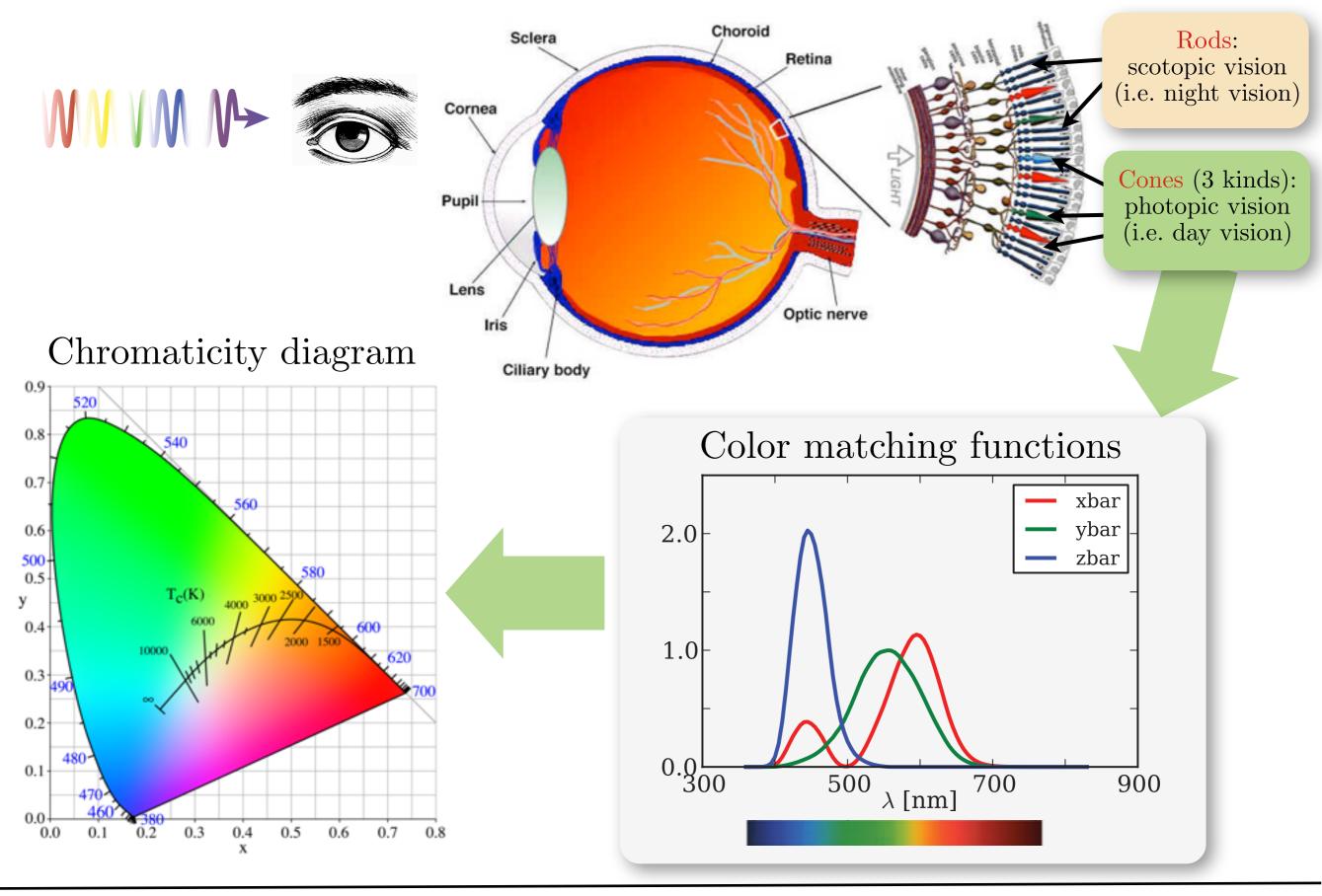




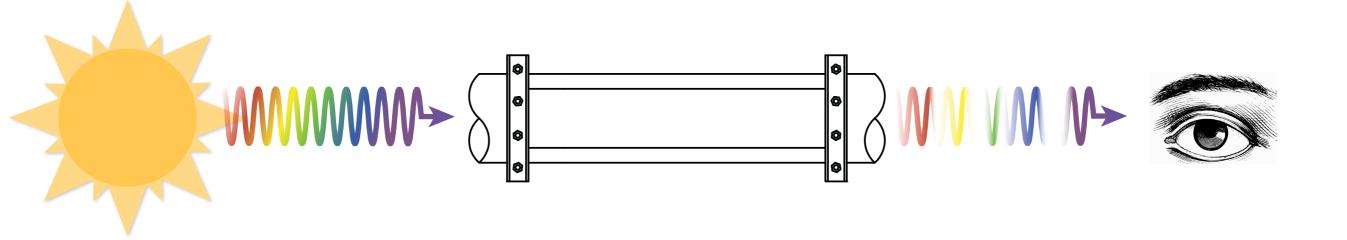


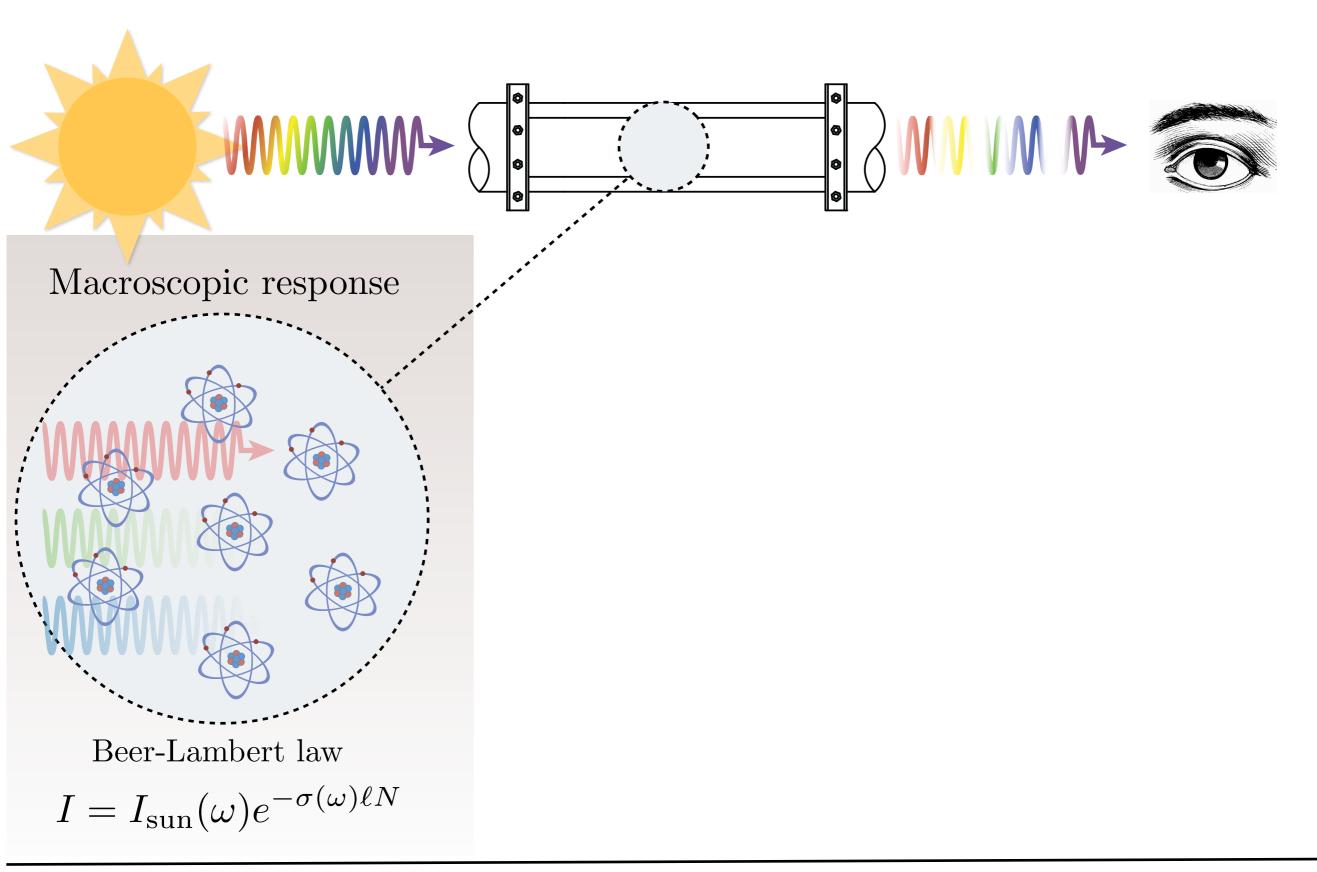


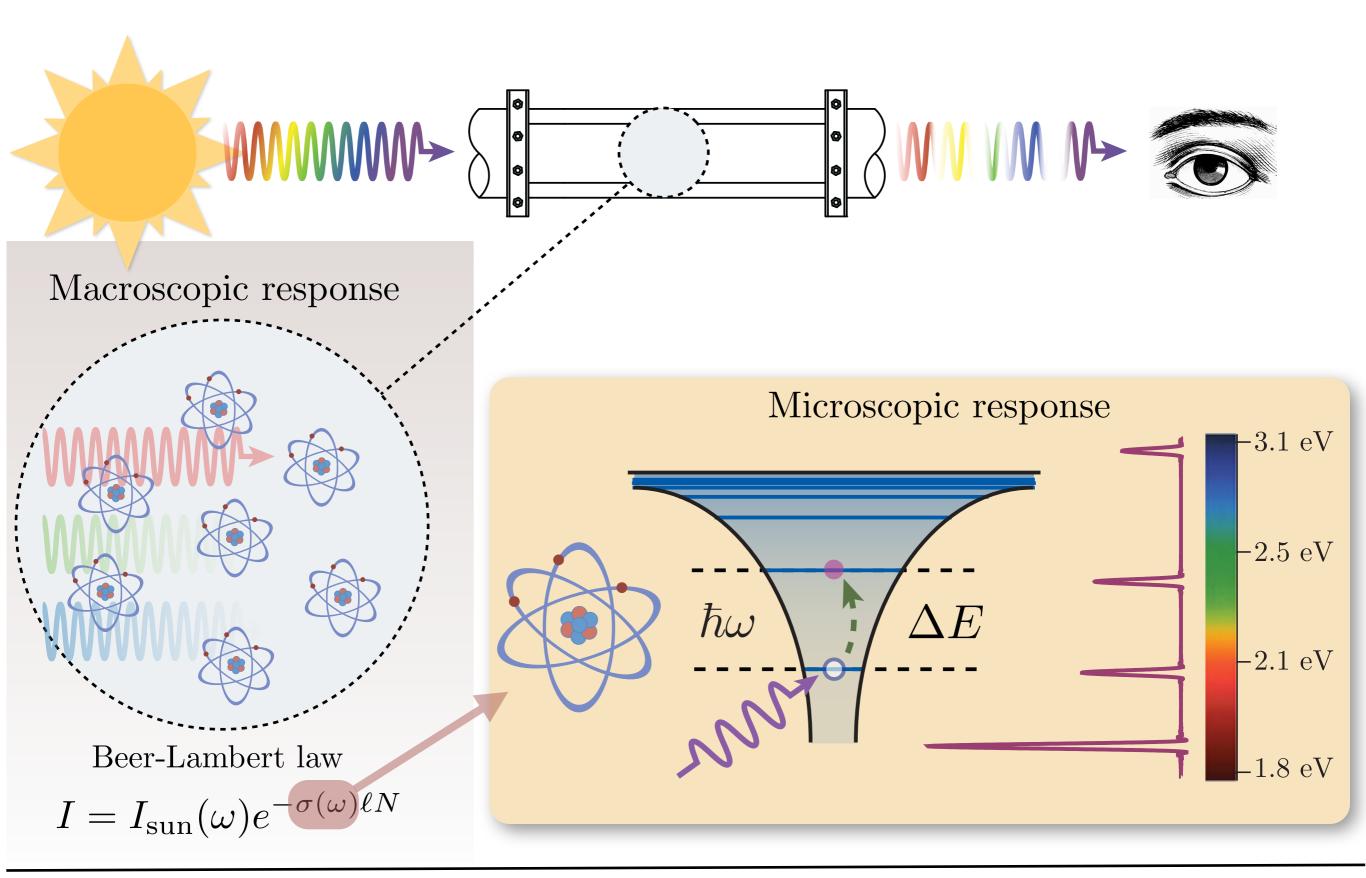


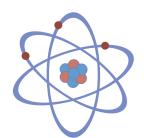


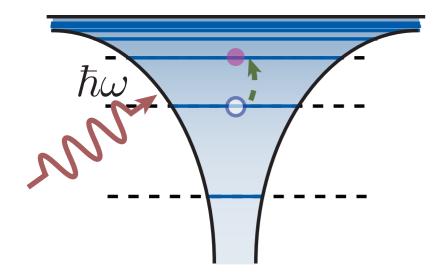
Santander September 23rd 2014

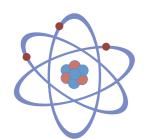


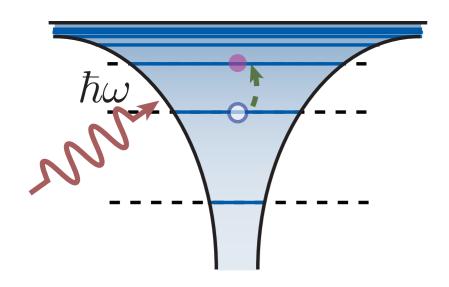






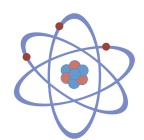


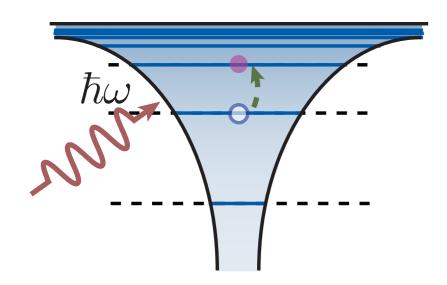




Hydrogen  

$$\Delta E(n, N) = E_0 \left[ \frac{1}{N^2} - \frac{1}{n^2} \right]$$
N=1 Lyman series (UV)  
N=2 Balmer series (VIS)  
N=3 Paschen series (IR)



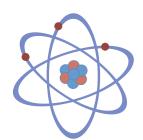


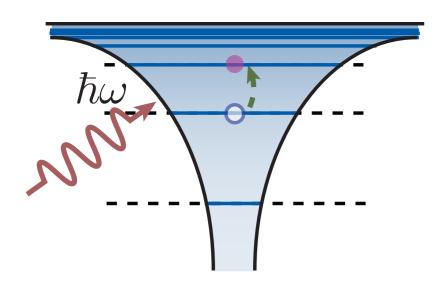
Hydrogen  

$$\Delta E(n, N) = E_0 \left[ \frac{1}{N^2} - \frac{1}{n^2} \right]$$
N=1 Lyman series (UV)  
N=2 Balmer series (VIS)  
N=3 Paschen series (IR)

 Larger systems have more complex excited-state structure

 Image: structure

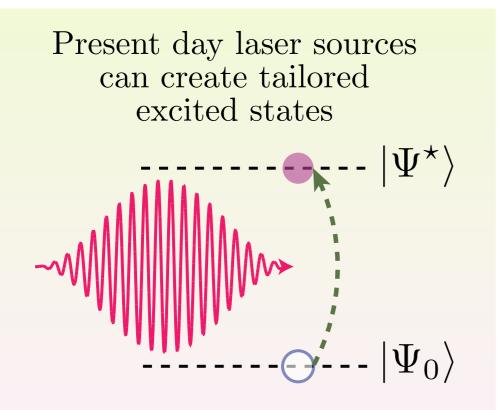


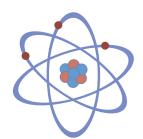


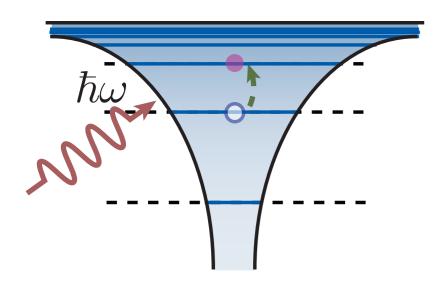
Hydrogen  

$$\Delta E(n, N) = E_0 \left[ \frac{1}{N^2} - \frac{1}{n^2} \right]$$
N=1 Lyman series (UV)  
N=2 Balmer series (VIS)  
N=3 Paschen series (IR)

Larger systems have more complex excited-state structure





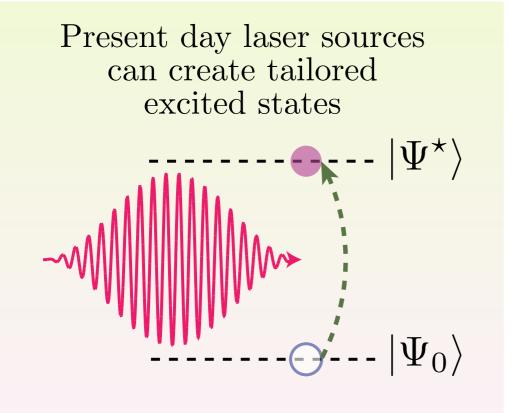


Hydrogen  

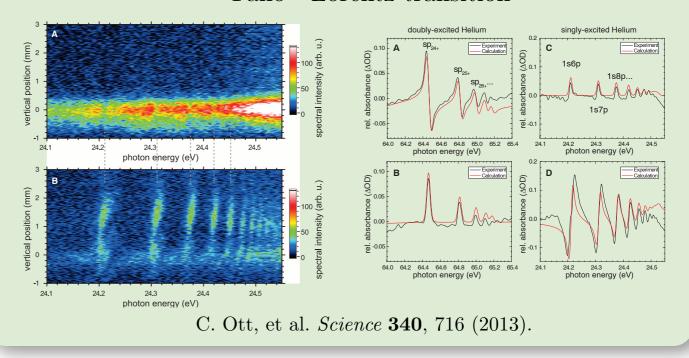
$$\Delta E(n, N) = E_0 \left[ \frac{1}{N^2} - \frac{1}{n^2} \right]$$
N=1 Lyman series (UV)  
N=2 Balmer series (VIS)  
N=3 Paschen series (IR)

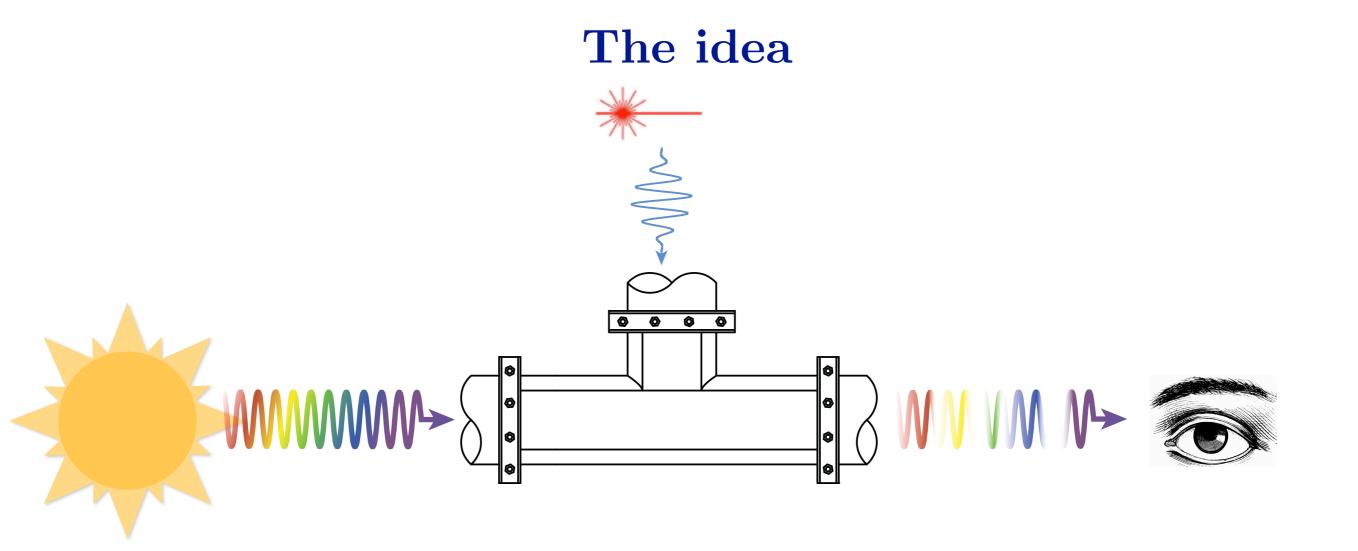
 Larger systems have more complex excited-state structure

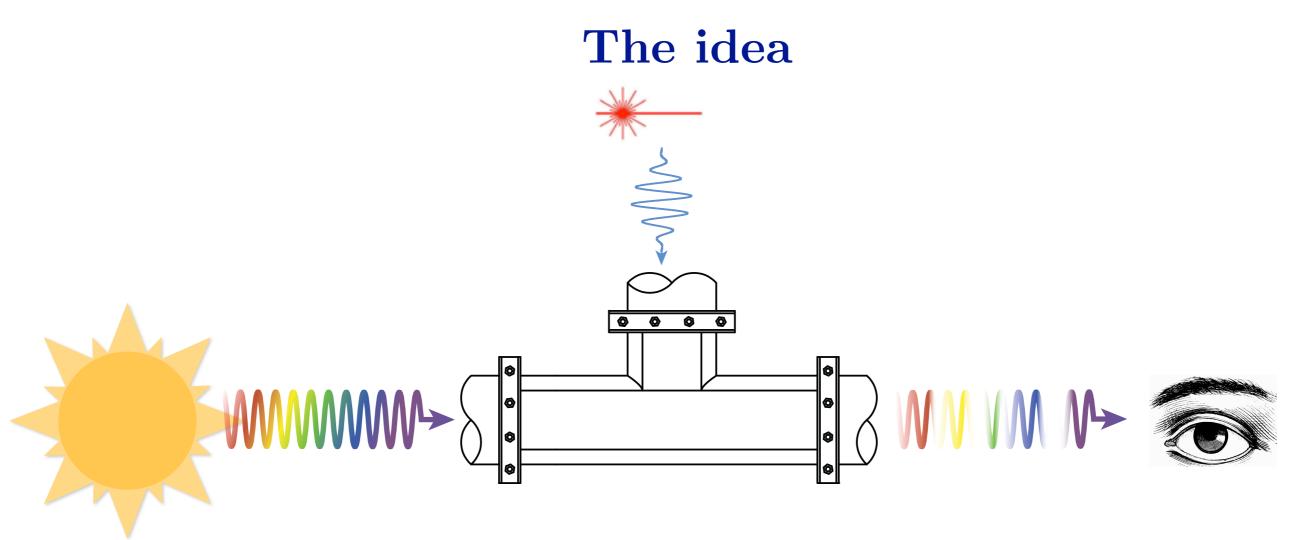
 Image: structure



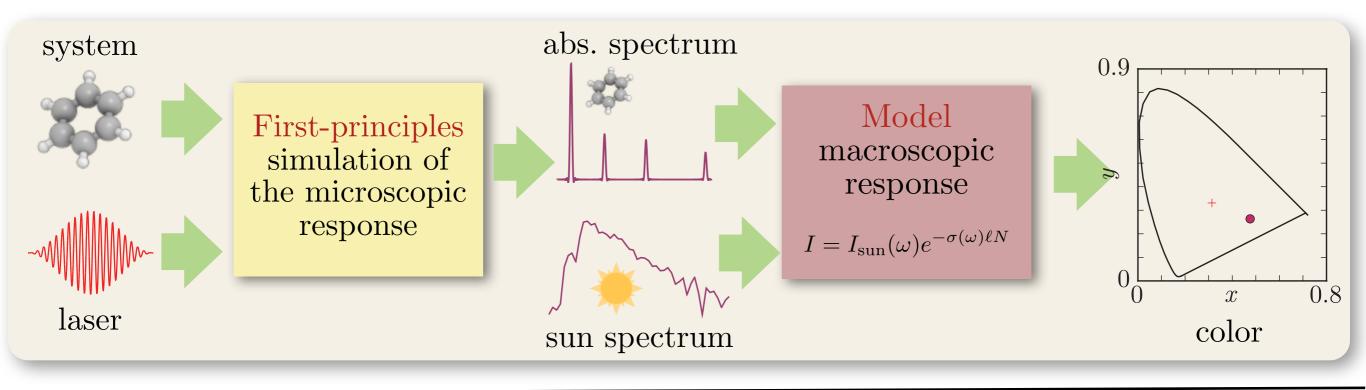
**In principle we can do more** Fano - Lorentz transition

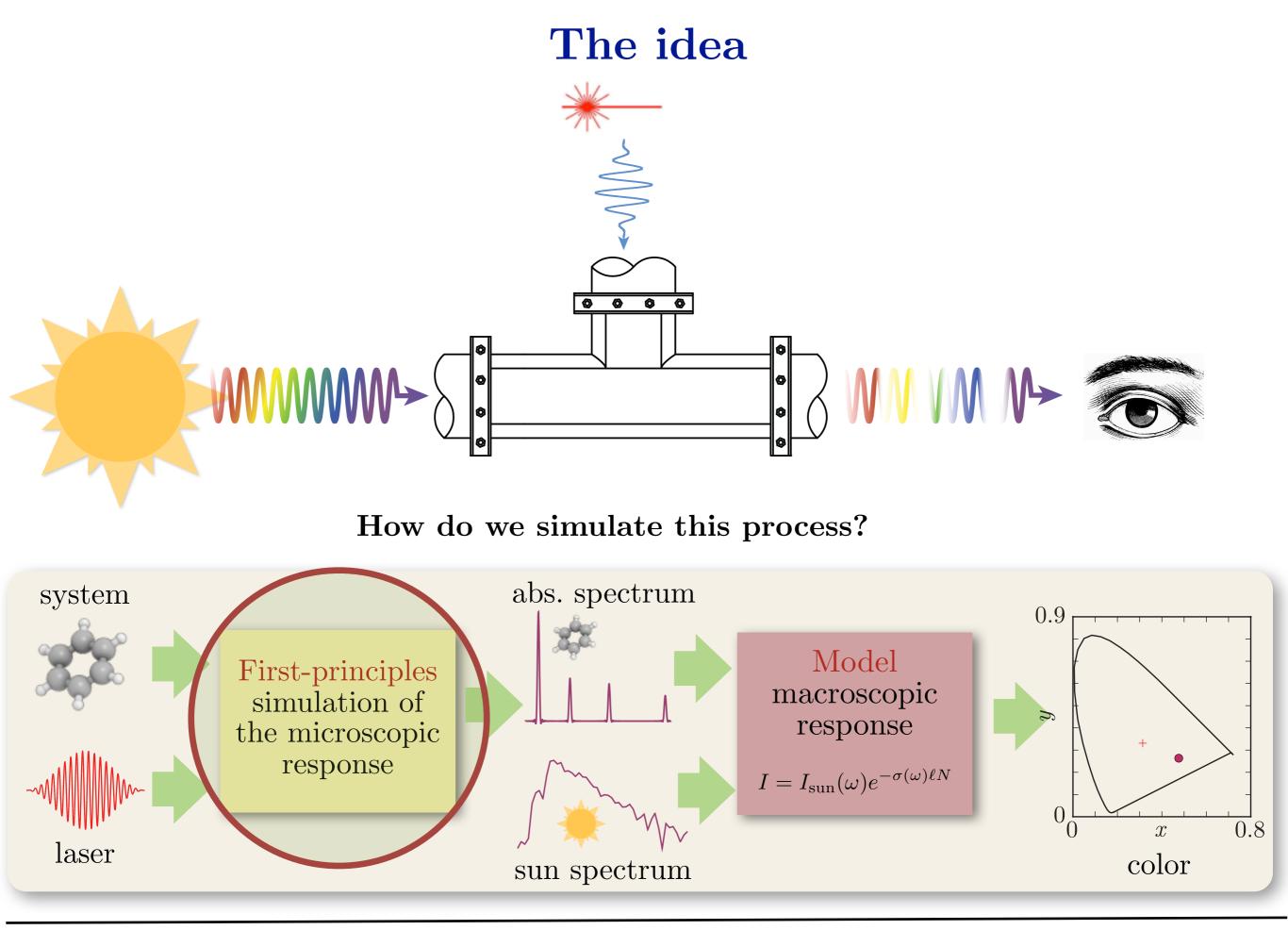






How do we simulate this process?





Time-dependent density functional theory

#### OCTOPUS

real-space real-time FD

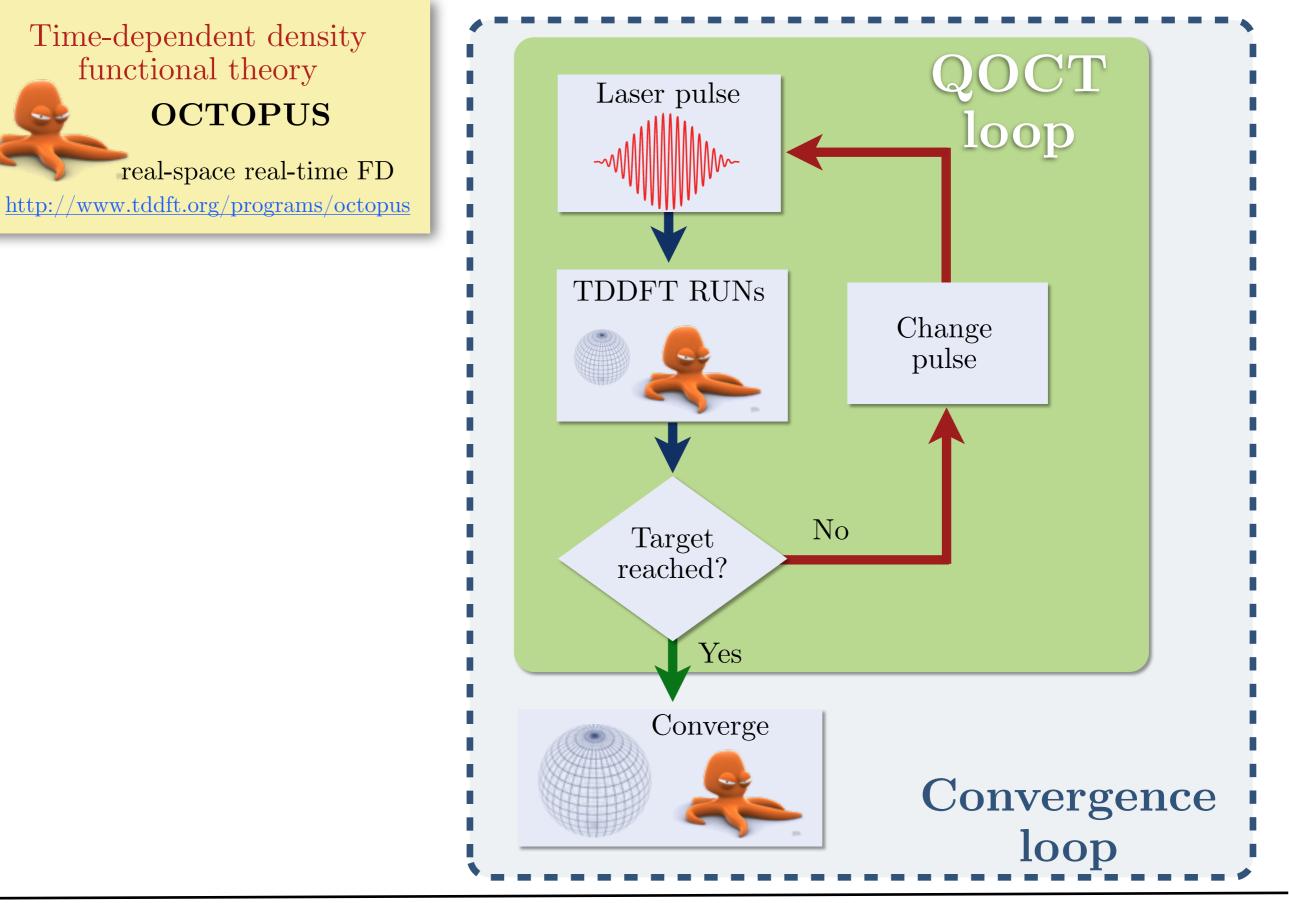
http://www.tddft.org/programs/octopus

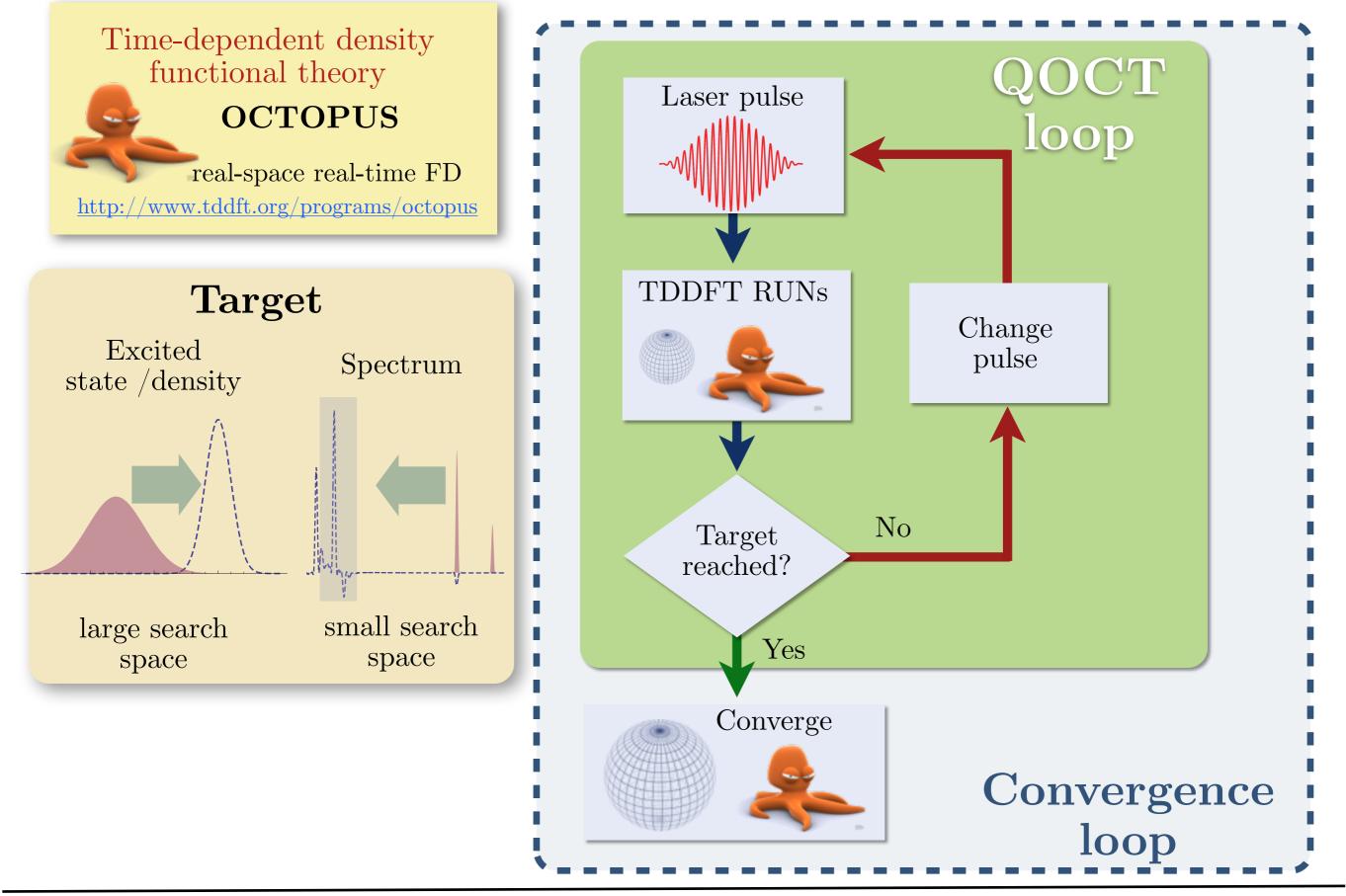
Time-dependent density functional theory

OCTOPUS

real-space real-time FD

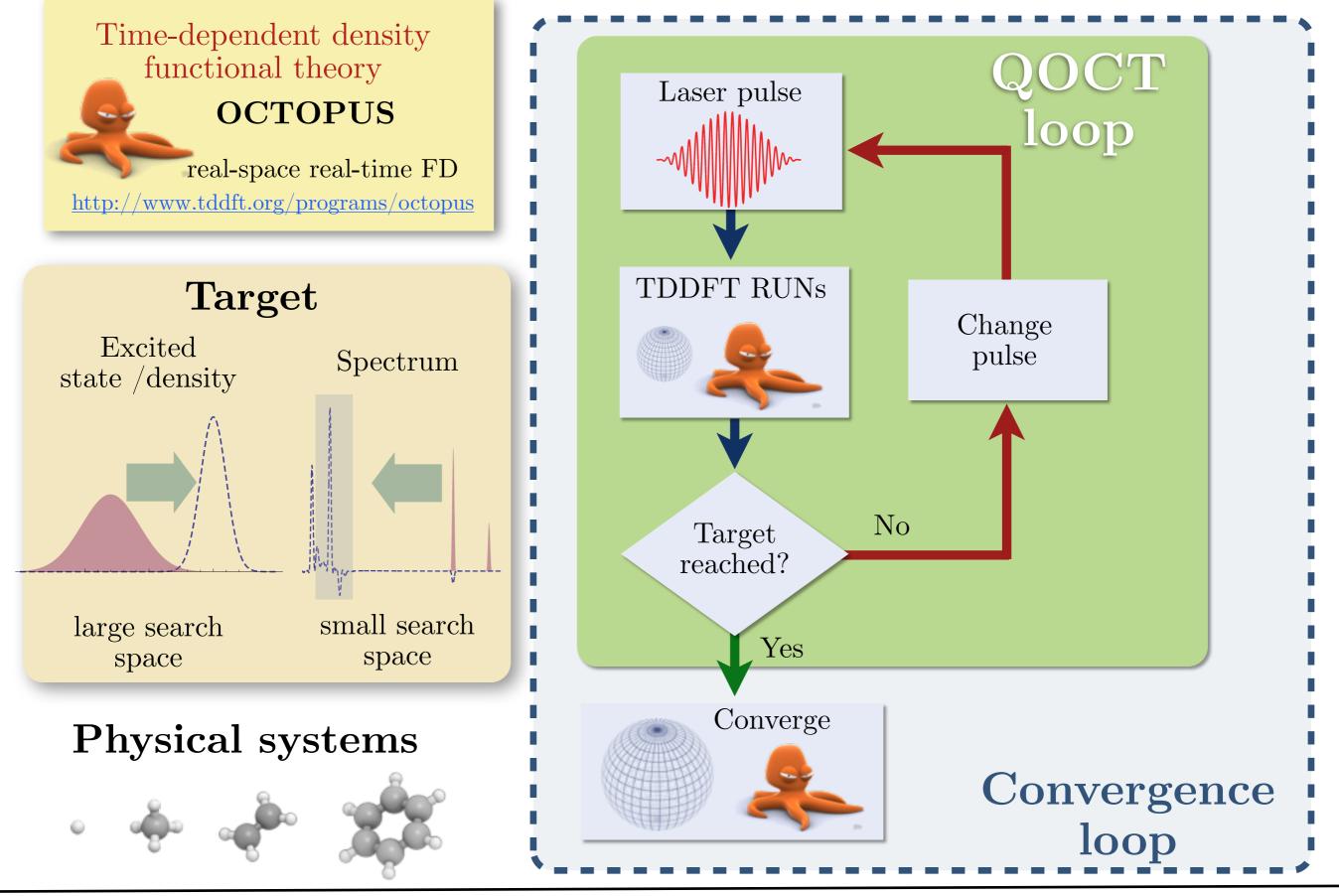
http://www.tddft.org/programs/octopus





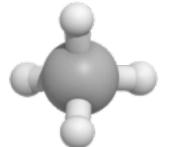
Santander September 23rd 2014

Theory & Tools

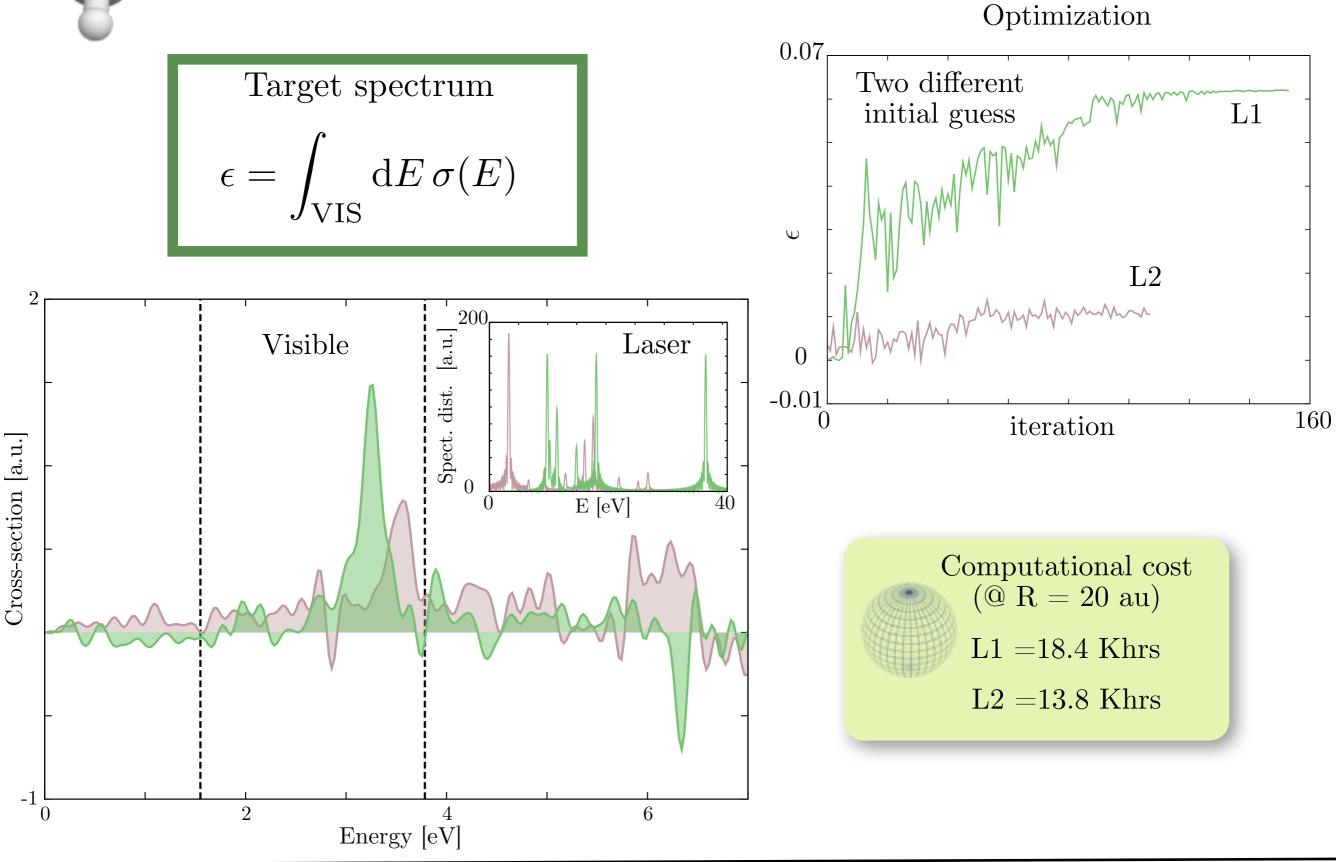


Santander September 23rd 2014

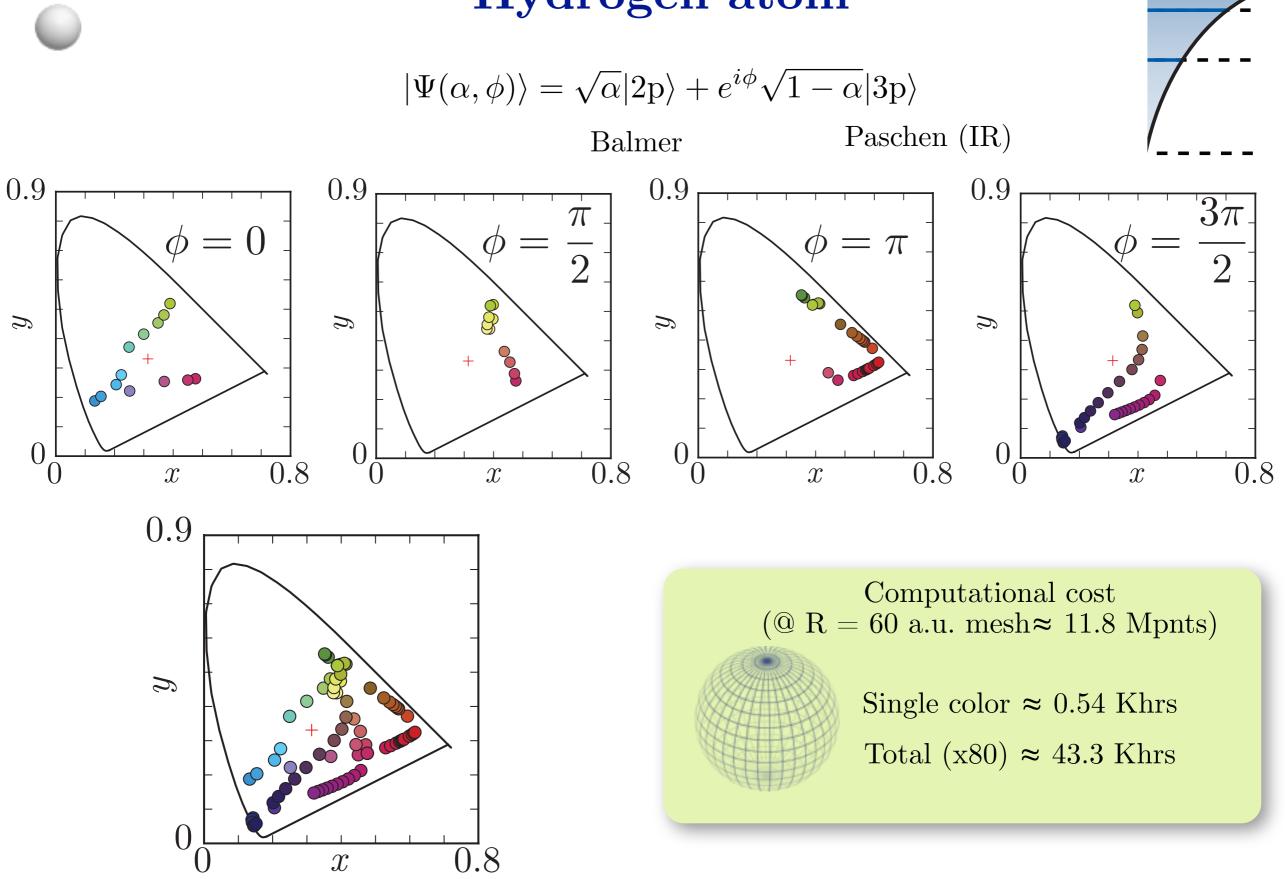
Theory & Tools



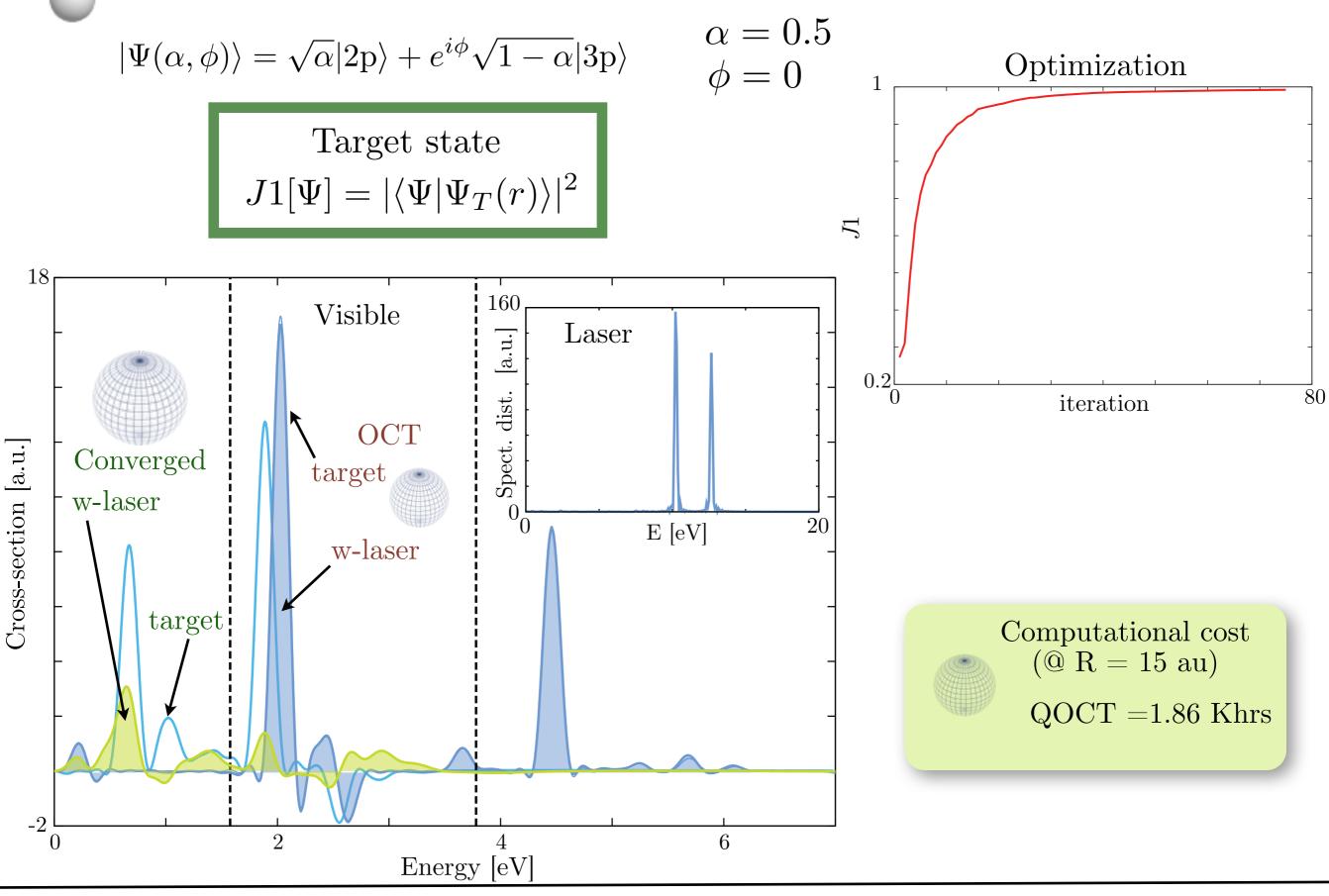
## Doubly ionized methane(CH4-2e)



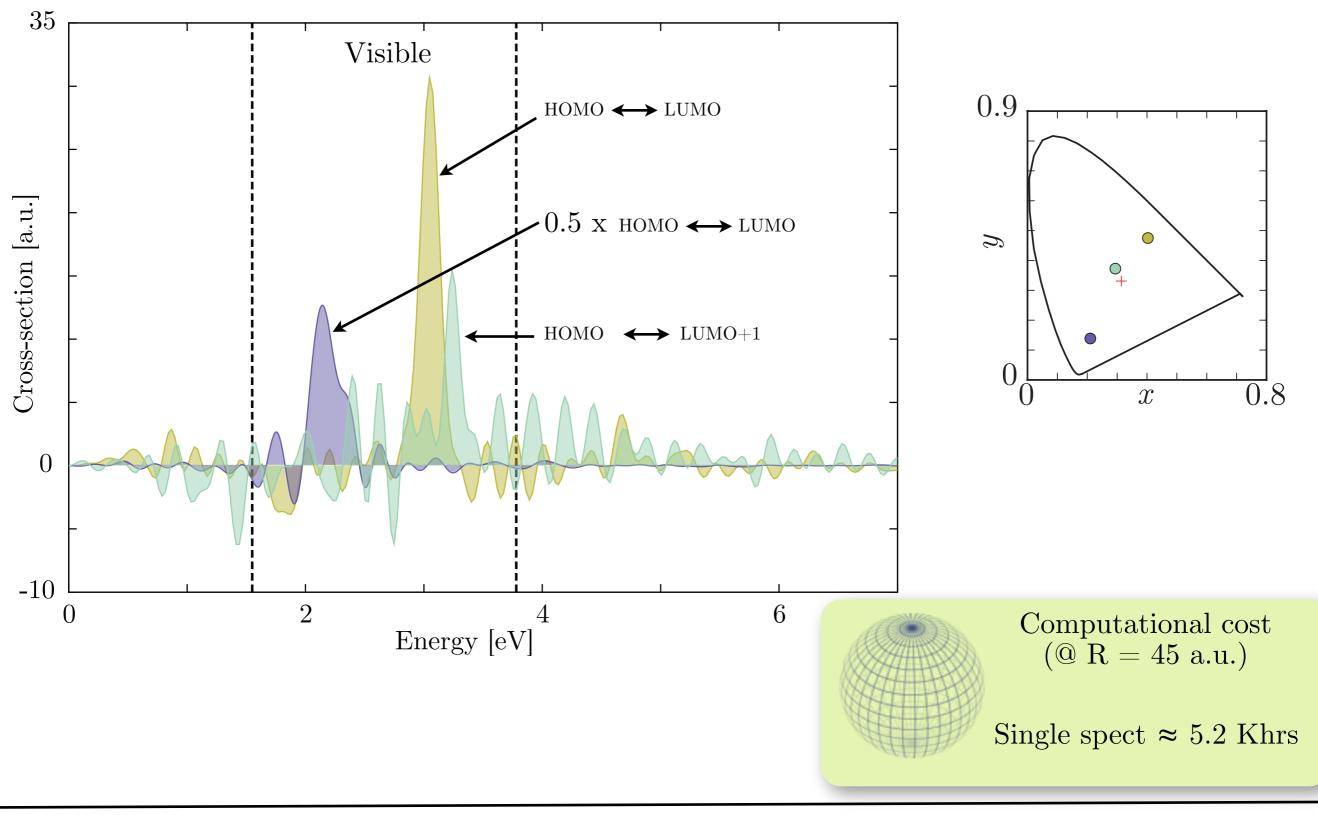
#### Hydrogen atom



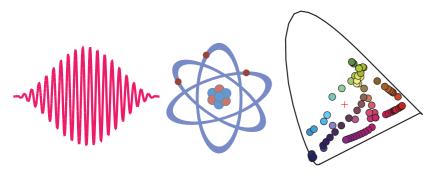
#### Hydrogen atom QOCT



## Methane (CH4)



Santander September 23rd 2014



#### Conclusions

RES allowed us to:

- test OCT strategies
- search for interesting target states
- identify exciting fields
- $\bullet$  investigate different systems

# THANK YOU!!!